JEE Main 2015 Chemistry



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16. Match the catalysts to the correct processes:

Catalyst Process

(A) TiCl₃ (i) Wacker process

(B) PdCl₂ (ii) Ziegler - Natta polymerization

(C) CuCl₂ (iii) Contact process (D) V₂O₅ (iv) Deacon's process

(1) $A \rightarrow (iii)$, $B \rightarrow (ii)$, $C \rightarrow (iv)$, $D \rightarrow (i)$ (2) $A \rightarrow (ii)$, $B \rightarrow (i)$, $C \rightarrow (iv)$, $D \rightarrow (ii)$ (3) $A \rightarrow (iii)$, $B \rightarrow (iii)$, $C \rightarrow (iv)$, $D \rightarrow (iv)$

Answer:

- (A) $TiCl_3 \rightarrow Ziegler$ -Natta polymerisation (Catalyst for ploymerisation ethylene and olefins). Ziegler-Natta catalysts are used to polymerize terminal 1-alkenes (ethylene and alkenes with the vinyl double bond)
- (B) $V_2O_5 \rightarrow$ Contact process (for producing Sulphuric acid)
- (C) PdCl₂ →Wacker process (an industrial process for the manufacture of ethanol by oxidizing ethene)
- (D) CuCl₂→ Deacon's process(The reaction takes place at about 400 to 450 °C in the presence of catalysts copper chloride (CuCl2) for producing chlorine gas from hydrochloride)

We know that in contact process of making sulphuric acid vanadium pentaoxide (V_2O_5) is used in given option D \rightarrow (iii) only in option (2),

so correct choice is option (2)