JEE Main 2015 Chemistry



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24. The synthesis of alkyl fluorides is best accomplished by:

(1) Free radical fluorination

(2) Sandmeyer's reaction

(3) Finkelstein reaction

(4) Swarts reaction

Answer:

Free radical fluorination: This chemical reaction is typical of alkanes and alkyl-substituted aromatics under application of UV light. The reaction is used for the industrial synthesis of chloroform (CHCl3), dichloromethane (CH2Cl2), and hexachlorobutadiene. It proceeds by a free-radical chain mechanism.

Sandmeyer's reaction: The Sandmeyer reaction is a chemical reaction used to synthesize aryl halides from aryl diazonium salts.

Finkelstein reaction: Treatment of a primary alkyl halide or pseudohalide with an alkali metal halide (e.g. KF, KI) leads to replacement of the halogen via an nucleophilic substitution ($S_N 2$ reaction).

Swarts reaction: The reaction of chlorinated hydrocarbons with metallic fluorides to form chlorofluorohydrocarbons. So alkyl fluorides synthesis is best accomplished by Swarts reaction.

$$R-X + AgF \rightarrow R-F + AgX$$

Thus alkyl fluoride is best accomplished by Swarts reaction.

Correct option is (4) Swarts reaction